This article was downloaded by:

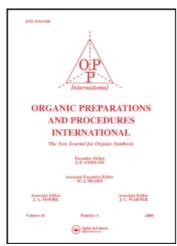
On: 27 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



## Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t902189982

# A NEW SYNTHESIS OF 2,5-DI-*t*-BUTYL-5,6-DICHLORO-1,4-CYCLOHEXENEDIOINE

Mircea D. Gheorghiu<sup>a</sup>; Margareta Avram<sup>a</sup>
<sup>a</sup> Institute of Organic Chemistry, Bucharest, Rumania

**To cite this Article** Gheorghiu, Mircea D. and Avram, Margareta(1974) 'A NEW SYNTHESIS OF 2,5-DI-*t*-BUTYL-5,6-DICHLORO-1,4-CYCLOHEXENEDIOINE', Organic Preparations and Procedures International, 6: 5, 231 — 233

To link to this Article: DOI: 10.1080/00304947409355110 URL: http://dx.doi.org/10.1080/00304947409355110

### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

# A NEW SYNTHESIS OF 2,5-DI-t-BUTYL-5,6-DICHLORO-1,4-CYCLOHEXENEDIONE

Mircea D. Gheorghiu and Margareta Avram X

Institute of Organic Chemistry, Spl. Independenței 202B

Bucharest 15, Rumania

The remarkable reactivity of <u>t</u>-butylcyanoketene (VI) in <u>2 + 2</u> cycloadditions prompted us to investigate its reactions with acetylenes and olefins. The starting material for the preparation of VI is compound V, obtained from 2,5-di-t-butylbenzoquinone (II) <u>via</u> alternative chlorination and dehydrochlorination to IV followed by treatment with sodium azide. <sup>3</sup> We now report a modified preparation of interme-

@ 1974 by Organic Preparations and Procedures, Inc.

#### M. D. GHEORGHIU AND M. AVRAM

diate III directly from readily accessible 2,5-di- $\underline{t}$ -butyl-hydroquinone (I), <sup>4</sup> thus avoiding the oxidation step to II.

The title compound (III) was obtained very pure and in high yield (9%) by chlorination in either technical grade or glacial acetic acid, during 3-4 hours, depending on the chlorine flow and the quantity of 2,5-di-t-butylhydroquinone to be chlorinated. Although in this oxidation-addition reaction, hydrogen chloride is evolved, no de-t-butylation occurred, unlike the observation of Moore 5 for related systems but in anhydrous conditions.

To our best knowledge, this conversion of a hydroquinone derivative to a 5,6-dichloro-1,4-cyclohexenedione is the first example reported until now in the literature.

#### EXPERIMENTAL

2.5-Di-t-butyl-5.6-dichloro-1.4-cyclohexenedione (III). Chlorine was bubbled through a well-stirred suspension of 60 g (0.27 mol) of 2,5-di-t-butylhydroquinone (I) 4 in 450 ml (90-100%) acetic acid at room temperature. After a few minutes, the reaction mixture became yellow, then dark (due to formation of quinhydrone) then pale yellow after 3-4 hours. When the whole suspension had this colour, the reaction was considered completed. The excess of chlorine was removed under reduced pressure with stirring. The suspension filtered, the filtrate poured into 500 ml cold water and refiltered. The combined solid was washed with chilled water, then dried in vacuo. The yield of 2,5-di-t-butyl-5,6-dichloro-1,4-cyclohexenedione (III) was 74.7 g (95%), mp. 127° (crude material, analytically pure), lit. 3 127-8°.

## 2,5-DI-t-BUTYL-5,6-DICHLORO-1,4-CYCLOHEXENEDIONE

This compound was converted into 2,5-di- $\underline{t}$ -butyl-3,6-dichlorobenzoquinone (IV). 3

#### REFERENCES

- 1. M. D. Gheorghiu, L. Stănescu, C. Drăghici and M. Avram, Tetrahedron Letters, 1973, 9.
- 2. M. D. Gheorghiu, F. Kerek and M. Avram, Rev. Roumaine Chim., 20 (1975), in press.
- H. W. Moore and W. Weyler, Jr., J. Am. Chem. Soc., 93,
   2812 (1971); J. Wikholm and H. W. Moore, ibid., 94,
   6152 (1972).
- 4. P. F. Oesper, C. P. Smyth and M. S. Kharash, J. Am. Chem. Soc., <u>64</u>, 937 (1942).
- H. W. Moore, D. L. Maurer, D. S. Pearce and M. S. Lee,
   J. Org. Chem., 37, 1984 (1972).

(Received June 27, 1974; in revised form October 22, 1974)